

STUDIES OF ACENAPHTHENE DERIVATIVES—XXVIII¹

THE REACTION OF 2-DIAZOACENAPHTHENONE WITH ANILS IN THE PRESENCE OF BORON TRIFLUORIDE

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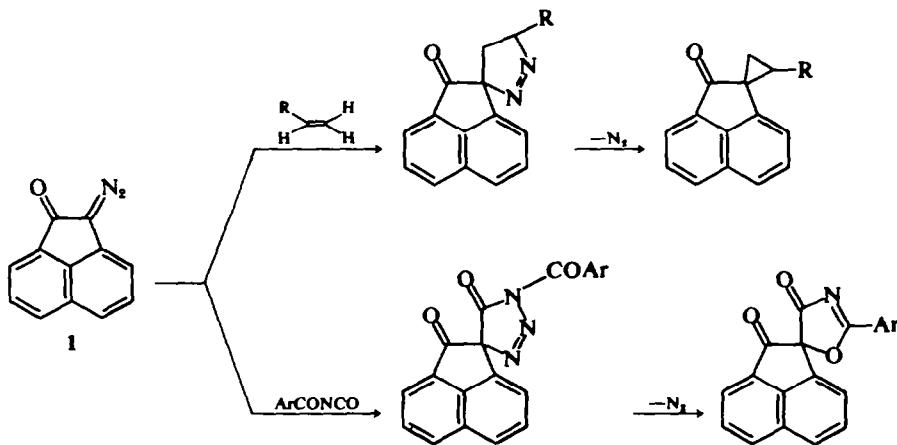
Abstract—Reaction of 2-diazoacenaphthenone with anils in the presence of a catalytic amount of boron trifluoride etherate is reported. The diazoketone reacts with benzyldenylanilines to afford a mixture of stereoisomers of spiro[acenaphthenone-2',4-imidazolidine], whose structures correspond to compounds derived from 1:2 adducts of the diazoketone and anils with the elimination of nitrogen. A pathway involving spiro-aziridine intermediates is proposed for their formation. Reaction of the diazoketone with α -alkylbenzyldenylanilines however gives the corresponding acenaphtho[1,2-b]pyrroles, whose structures correspond to compounds arising from 1:1 adducts with the elimination of both nitrogen and water. The reaction of the tautomeric enamine form of the anil with a carbenoid intermediate generated from the diazoketone is suggested for pyrrole formation.

In earlier publications, we have shown that the reactions of 2-diazoacenaphthenone (1) with olefins² and with aroyl isocyanates³ under mild conditions afforded spiro-cyclopropanes and spiro-oxazolidinones respectively. These reactions can be viewed as proceeding *via* a 1,3-cycloaddition of the diazomethane moiety of 1 to the unsaturated bonds, followed by the elimination of nitrogen and concurrent recyclization as shown in Scheme 1.

Thus, in the cycloaddition reaction not involving the ketocarbene, 1 behaves exclusively as a 1,3-dipole rather than as a 1,5-dipole.

Although the addition of diazoalkanes to activated olefinic double bonds is a well known route to the preparation of pyrazolines,⁴ few reports are available on the addition of diazoalkanes to carbon-nitrogen double bonds. Mustafa⁵ observed the formation in small quantities of 1,2,3-triazolines by the 1,3-cycloaddition of diazomethane to anils, and solvent effects on the cycloaddition have been investigated by Kadaba.⁶ However, there is little information about the reaction of α -diazoketones with anils. If the addition of 1 to anils would take place in a similar manner to olefins, spiro-aziridines would be the expected products. It thus seemed of interest to investigate the reaction of 1 with anils.

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SCHEME 1

RESULTS AND DISCUSSION

Reaction with benzyldenylanilines. 2-Diazoacenaphthenone (1) did not react with benzyldenylanilines (2) upon prolonged reflux in benzene and was recovered in a quantitative yield.

When a benzene solution of 1 and *p*-chlorobenzylidene-*p*-chloroaniline (2a) was refluxed in the presence of catalytic amounts of boron-trifluoride etherate (BF₃·Et₂O) for 1 h, a crystalline compound 3a (m.p. 232–234°) was obtained in 33.8% yield, accompanied by trace amounts of 2,2'-diacenaphthylidene-1,1'-dioxide (biacenedione); 3a was determined to be a 1:1.5:1 mixture of three isomeric compounds, 3a-1 (m.p. 234–235°), 3a-2 (m.p. 241–242°) and 3a-3 (m.p. 273–274° dec), all as yellow prisms.

Microanalyses and molecular ion (M⁺ *m/e* 666) of all three isomers fitted the molecular formula of a compound arising from the elimination of nitrogen from a 1:2 adduct of 1 and 2a. The IR spectra of 3a-1, 3a-2 and 3a-3 showed bands ascribable to ν_{C=O} at 1720, 1730 and 1732 cm⁻¹ respectively and did not exhibit any ν_{NH} absorption.

The NMR spectrum of 3a showed three singlets at δ 5.43, 5.19 and 5.52 ppm in a 2:3:2 ratio; these singlets appeared as 1H in the spectra of pure samples of 3a-1, 3a-2 and 3a-3. Furthermore, the mass spectra of all the isomers were identical and displayed fragment peaks at *m/e* 415 (M⁺-2a,

46.5%) and 290 (415⁺-Cl-C₆H₄-N, 23%) besides

a parent ion (M⁺) at *m/e* 666. On the basis of these observations, it was deduced that 3a-1, 3a-2 and 3a-3 were stereoisomers of a 1,2,3,5-tetrakis(*p*-chlorophenyl)-spiro[acenaphthenone-2',4'-imidazolidine], although their stereochemistry is not clear in the present stage.

Under similar conditions, 1 was reacted with benzyldenylaniline (2b), *p*-methylbenzyldenylaniline (2c) and *p*-chlorobenzylidene-*p*-toluidine (2d) to afford the corresponding spiro[acenaphthenone-

2',4'-imidazolidine] derivatives 3b, 3c and 3d. The NMR spectra indicated that 3b, 3c and 3d were also mixtures of stereoisomeric spiro-imidazolidines, but it was not possible to separate the respective isomers. The yields, physical properties and microanalyses of 3 are summarized in Table 1.

Hydrolysis of 3a and 3d with conc. HCl in refluxing EtOH gave the corresponding 2-ethoxy-2-(α -ethoxybenzyl)acenaphthenone 4a, and *p*-chloroaniline and *p*-toluidine respectively. Similar treatment of 3a in refluxing MeOH afforded 2-methoxy-2-(α -methoxybenzyl)acenaphthenone 4b and *p*-chloroaniline. The structures of 4a and 4b were confirmed by their spectral data and microanalyses.

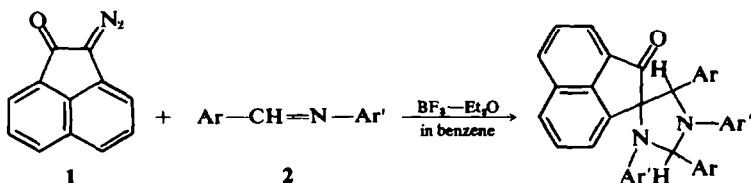
However on treatment of 3d with conc. HCl at room temp, acenaphthenequinone mono-*p*-toluidine (5), *p*-chlorobenzaldehyde and *p*-toluidine were formed in addition to trace amounts of anil 2d (Scheme 2). These facts also support the proposed structures for 3.

Reaction pathway. It is known that the decomposition of diazoalkanes and α -diazocarbonyl compounds is catalyzed by boron trihalides. In the case of CH₂N₂, the first step is without doubt coordination of the Lewis acid with the carbon atom, but there is some question about the subsequent steps leading to polymethylene.⁷ On the other hand, it is considered that boron trihalide coordinates with the oxygen atom of the carbonyl group in α -diazoketones.⁸

Treatment of 1 with BF₃·Et₂O in benzene at room temp gave extremely unstable yellow crystals (possibly a complex) which could not be isolated† but when a benzene solution of anil 2a was refluxed with a catalytic amount of BF₃·Et₂O, 2a was recovered in 84% yield. Thus it appears that BF₃ reacts preferentially with 1 rather than with 2 in the reaction of 1 with 2 in the presence of BF₃·Et₂O.

We view the pathway for the formation of spiro[acenaphthenone-2',4'-imidazolidine] derivatives 3 as depicted in Scheme 3. Although the postulation of C-bonded complex A is unusual, our previous results^{2,3} suggested that the contribution of the mesomeric 1,5-dipolar form to the structure of 1 is small. The formation of a reactive species B (an "inverse ylid") may proceed either directly *via* A

†While decomposition of 1 with catalytic amounts of BF₃·Et₂O in refluxing benzene gave 2,2'-diphenyl-2,2'-dihydrodiacenaphthylidene-1,1'-dione, with excess of BF₃·Et₂O 2-phenylacenaphthenone was obtained. However, reaction pathways are not clear.



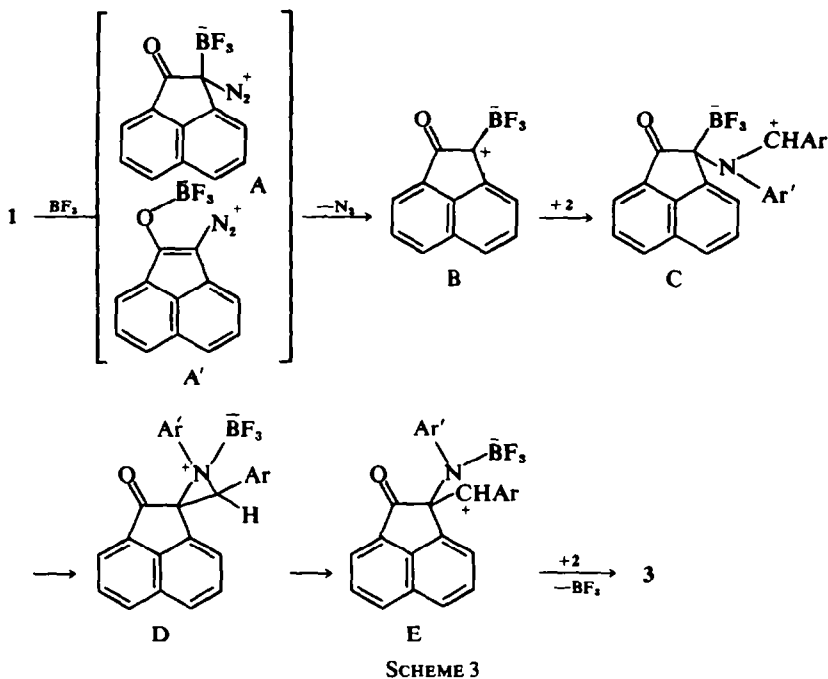
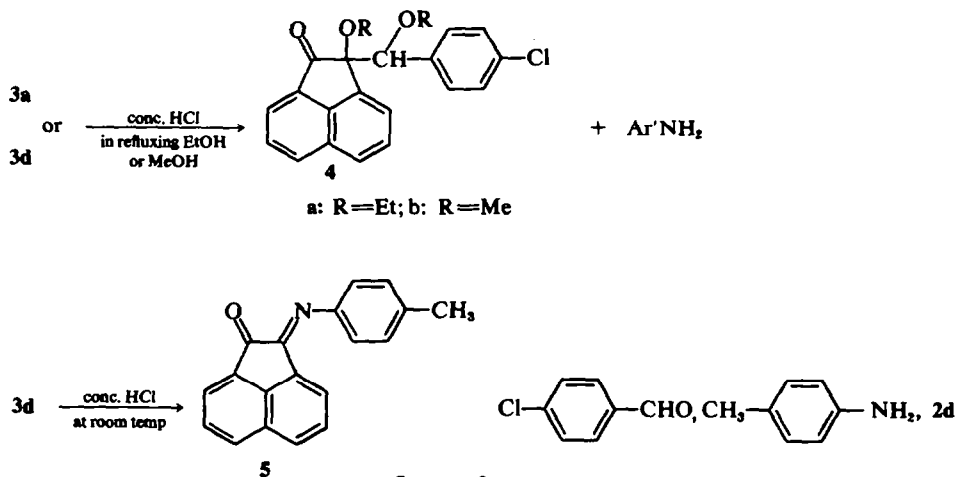
- a: Ar=Ar'=*p*-ClC₆H₄; b: Ar=Ar'=Ph
 c: Ar=*p*-CH₃C₆H₄, Ar'=Ph
 d: Ar=*p*-ClC₆H₄, Ar'=*p*-CH₃C₆H₄

Table 1. Spiro[acenaphthenone-2',4'-imidazolidines] 3

3 ^a	Ar	Ar'	Reaction time (min)	Yield (%)	M.p. (°C)	IR ν_{CO} (cm ⁻¹)	NMR(CDCl ₃) ^b CH δ ppm (rel. ratio)	Microanalysis (%)			M ⁺ (m/e)
								Found	Calc.	Found	
3a-1	<i>p</i> -ClC ₆ H ₄	<i>p</i> -ClC ₆ H ₄		234-235	1720	5.43	68.80 (68.47)	3.47 (3.60)	4.08 (4.20)	666	
3a-2	"	"	60	33.8	241-242	1730	68.71 (68.47)	3.58 (3.60)	4.16 (4.20)	666	
3a-3	"	"			273-275 dec	1732	68.50 (68.47)	3.49 (3.60)	4.22 (4.20)	666	
3b	C ₆ H ₅	C ₆ H ₅	40	30.8	248-248.5	1718	86.54 (86.33)	5.02 (5.34)	5.30 (5.30)	528	
3c	<i>p</i> -CH ₃ C ₆ H ₄	C ₆ H ₅	10	31.4	136-138	1726	86.59 (86.30)	5.62 (5.79)	4.78 (5.03)	556	
3d	<i>p</i> -ClC ₆ H ₄	<i>p</i> -CH ₃ C ₆ H ₄	30	32.2	157-160	1728	76.75 (76.80)	4.72 (4.80)	4.71 (4.48)	625	

^aAppearance: 3a-1 – 3a-3, yellow prisms; 3b, pale yellow prisms (MeOH); 3c, light yellow grains (EtOH-H₂O); 3d, yellow prisms (EtOH-H₂O).

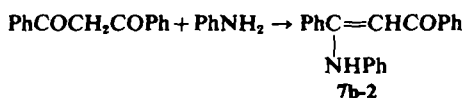
^bComplicated signals appeared in the spectra, besides the singlets indicated in the Table: 3a-1, δ 5.85-8.1; 3a-2, δ 6.1-8.2; 3a-3, δ 6.1-8.2 (each 23H); 3b, δ 6.0-8.1 (27H), 3c, δ 1.7-2.5 (6H, CH₃), 5.8-8.3 (25H); 3d, δ 1.9-2.2 (6H, CH₃), 5.8-8.1 (23H).



or *via* an initial O-bonded complex A' which may then go to reactive species B as shown in Scheme 3. The reactive species B would then react with the nitrogen atom of anil 2 to yield betaine intermediate C which would cyclize to spiroaziridine D, followed by ring opening to 1,3-dipole E.⁹ Cycloaddition of E to 2 would yield the final product 3.

The proposed pathway involving aziridine intermediate D was supported by the following evidence. When a solution of diazoacetophenone (6) and anil 2b in benzene containing a catalytic amount of BF₃·Et₂O was refluxed, compound 7b (C₂₁H₁₇NO, M⁺ *m/e* 299) whose molecular formula agreed with that of a compound arising from a 1 : 1 adduct with

N₂ elimination, was obtained. On the basis of its spectral data, two structures, α - and β -anilino-benzylidenacetophenone 7b-1 and 7b-2 respectively are possible for 7b. However, we found that product 7b was different from authentic 7b-2, prepared from dibenzoylmethane and aniline.¹⁰ Consequently, 7b was deduced to be α -anilino-benzylidenacetophenone (7b-1). Similarly, reaction of 6 with 2a afforded α -*p*-chloroanilino-*p*-chlorobenzylidenacetophenone (7a-1).



The pathway for the formation of 7-1 can be rationalized as depicted in Scheme 4. The 1,3-dipole G arising from aziridine intermediate F rearranges easily into stable product 7-1 with the elimination of BF_3 and concurrent hydrogen shift, because G has a hydrogen atom on the central carbon atom. In contrast, species E does not have a hydrogen atom at the 2-position and behaves as a real 1,3-dipole.

Reaction with α -alkylbenzylidenanilines. As mentioned above, the reaction of diazoketone 1 with benzylidenanilines (2) afforded a mixture of stereoisomeric spiro[acenaphthenone-2',4-imidazolidine] compounds 3, whose stereochemistry could not be clarified. If diazoketone 1 reacts with α -alkylbenzylidenanilines (8) in a similar manner, it would be expected to form a 2,5-dialkyl-1,2,3,5-tetraarylspiro[acenaphthenone-2',4-imidazolidine] (9) whose stereochemistry would be easily elucidated when compared with those of 2,5-unsubstituted spiro compounds 3.

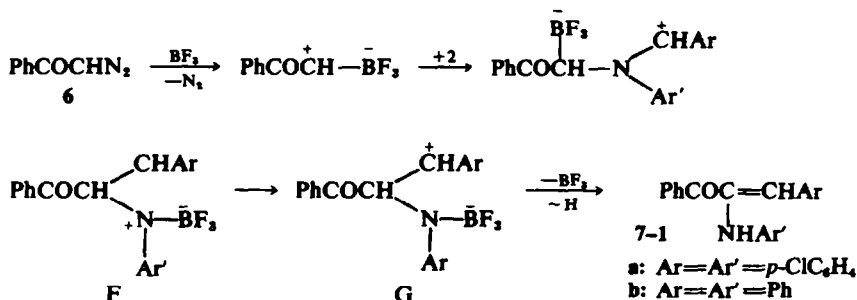
Contrary to expectation, the reaction of 1 with α -methyl- (8a) and α -ethylbenzylidenaniline (8b) afforded compounds 10a and 10b whose molecular formulas agreed with those of compounds derived

from the corresponding 1:1 adducts with the elimination of both nitrogen and water respectively. On the basis of their spectral data, these compounds were assigned as 1,2-diphenylacenaphtho[1,2-b]pyrrole (10a) and 1,2-diphenyl-3-methylacenaphtho[1,2-b]pyrrole (10b).

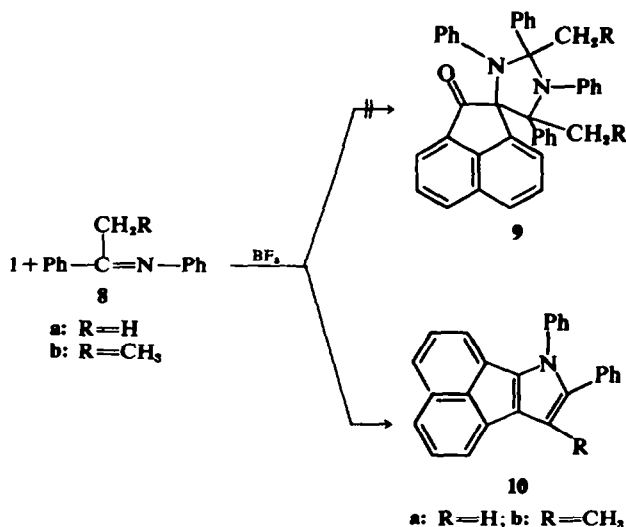
Although the exact pathway for the formation of 10 is not clear, we viewed the reaction as proceeding as shown in Scheme 5. It is well known¹¹ that α -alkylbenzylidenanilines (8) exist as their tautomeric enamines 8'. Therefore, reactive species B would attack the β -carbon atom of enamines 8' to form 1-hydroxy-2-(β -anilinostryl)acenaphthylene I via betaine H with the elimination of BF_3 and concurrent hydrogen shift. Intermediate I would then cyclize to the final product 10. In this case, an aziridine intermediate similar to D is not involved.

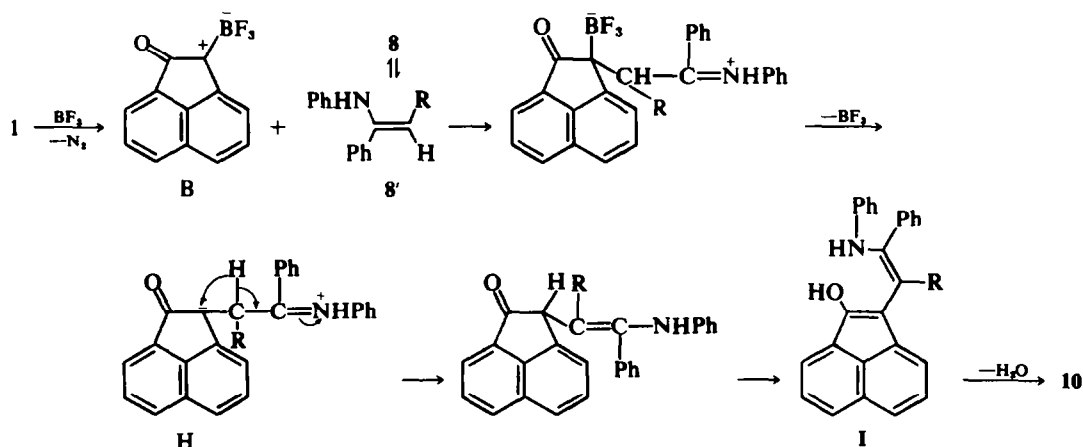
EXPERIMENTAL

M.ps and b.ps are uncorrected. IR spectra were measured as KBr discs and NMR spectra were determined at 60 MHz with a Hitachi R-20 NMR spectrometer (TMS as internal reference). Mass spectra were obtained on a Hitachi RMS-4 mass spectrometer (direct inlet; 70 eV), and microanalyses were performed by Miss M. Akita of our laboratory.



SCHEME 4





SCHEME 5

2-Diazoacenaphthenone (1) was prepared by the reported method¹² and purified by chromatography on alumina, m.p. 93–94° (lit.,¹² m.p. 94°). Diazoacetophenone (6) was prepared by reaction of benzoyl chloride and CH_3N_2 , m.p. 48–49° (lit.,¹³ m.p. 49°). All anils were prepared by literature methods. *p*-Chlorobenzylidene-*p*-chloroaniline (2a), m.p. 109–110° (lit.,⁶ m.p. 109–110.5°); benzylidenaniline (2b), m.p. 51–52° (lit.,⁶ m.p. 52°); *p*-methylbenzylidenaniline (2c), b.p. 170–173°/14 mm (lit.,¹⁴ b.p. 318°/760 mm); *p*-chlorobenzylidene-*p*-toluidine (2d), m.p. 124–125° (lit.,¹⁵ m.p. 125°); α -methylbenzylidenaniline (8a), b.p. 175°/15 mm (lit.,¹⁶ b.p. 170–180°/13 mm); α -ethylbenzylidenaniline (8b), b.p. 153°/5 mm (lit.,¹⁷ b.p. 169°/11 mm). $\text{BF}_3 \cdot \text{Et}_2\text{O}$ was purchased from Kishida Kagaku Co. (Osaka, Japan).

Decomposition of 1 in the presence of $\text{BF}_3 \cdot \text{Et}_2\text{O}$. (i) A solution of 400 mg (2 mmole) of 1 and 30 mg (0.2 mmole) of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in 10 ml benzene was refluxed for 1 h. After cooling, the mixture was chromatographed on alumina to give 20 mg of 2,2'-diphenyl-2,2'-dihydrodiacenaphthylidene-1,1'-dione, m.p. 245° dec, as colourless prisms. (Found: C, 89.02; H, 4.71. Calc. for $\text{C}_{36}\text{H}_{22}\text{O}_2$: C, 88.86; H, 4.56%). ν_{CO} 1720 cm^{-1} .

Further elution with CHCl_3 afforded trace amounts of biacenedione, m.p. 294°, which was identical with the authentic sample prepared from acenaphthenequinone and acenaphthenone.

(ii) Similarly, treatment of 400 mg of 1 with 1.5 g (10 mmole) of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in 20 ml of benzene gave 120 mg (27%) of 2-phenylacenaphthenone, m.p. 114–115° (lit.,¹⁸ m.p. 115–116°), as colourless grains. (Found: C, 88.70; H, 5.15. Calc. for $\text{C}_{18}\text{H}_{12}\text{O}$: C, 88.50; H, 4.95%). ν_{CO} 1716 cm^{-1} . δ ppm (CDCl_3): 4.86 (1H, s, \rightarrow CH), 7.5 (11H, m, aromatic protons).

Reaction of 1 with 2. The general procedure is illustrated by the reaction of 1 with 2a. A solution of 1.0 g (5 mmole) of 1 and 1.25 g (5 mmole) of 2a in 30 ml of benzene was refluxed for 1 h with 80 mg of $\text{BF}_3 \cdot \text{Et}_2\text{O}$. After cooling, the mixture was concentrated to 20 ml *in vacuo* and chromatographed on alumina to give 1.16 g (33.8%) of a mixture of spiro[acenaphthenone-2',4-imidazolidines] 3a, m.p. 232–234°, as yellow grains. (Found: C, 68.47; H, 3.62; N, 3.94. Calc. for $\text{C}_{38}\text{H}_{24}\text{N}_2\text{OCl}$: C, 68.47; H, 3.60; N, 4.20%). *m/e* (rel. intensity): 666 (trace), 417 (42.5), 416 (25), 415 (64.5), 291 (38), 290 (25), 289 (100).

Further elution with CHCl_3 afforded trace amounts of biacenedione, m.p. 294°.

Isolation of isomers from 3a was carried out by chromatography on silica gel using cyclohexane-benzene as eluent. From the elution with cyclohexane-benzene (2:1 vol/vol), 3a-1 and 3a-2 were obtained, and 3a-3 was isolated from the elution with cyclohexane-benzene (1:1 vol/vol). All isomers were recrystallized from EtOH.

Similarly, reaction of 1 with anils 2b, 2c and 2d afforded the corresponding spiro[acenaphthenone-2',4-imidazolidine] derivatives 3b, 3c and 3d. The yields, physical properties and microanalyses are in Table 1.

Hydrolysis of 3a. A solution of 400 mg of 3a in 18 ml of EtOH was refluxed with 2 ml of conc. HCl for 5 h. After solvent was evaporated residue was poured into water and extracted with benzene (20 ml \times 2). The extract was water washed, dried (Na_2SO_4) and concentrated to about 10 ml. The resulting solution was chromatographed on alumina (benzene) to give crystals, which on crystallization from petrol gave 90 mg (39.7%) of 2-ethoxy-2-(α -ethoxy-*p*-chlorobenzyl)acenaphthenone (4a), m.p. 139–140°, as colourless prisms. (Found: C, 72.49; H, 5.43. Calc. for $\text{C}_{23}\text{H}_{21}\text{O}_3\text{Cl}$: C, 72.54; H, 5.52%). ν_{CO} 1726 cm^{-1} . *m/e*: 379 (M^+), δ ppm (CCl_4): 0.80, 1.02 (each 3H, t, CH_3), 3.01, 3.23 (each 2H, q, CH_2), 4.87 (1H, s, \rightarrow CH), 7.2–8.2 (9H, m, aromatic protons).

The aqueous layer was neutralized with K_2CO_3 aq. and benzene extracted. The benzene was evaporated to leave 70 mg (45.5%) of *p*-chloroaniline.

Similarly, a solution of 1.0 g of 3a in 30 ml of MeOH was refluxed with 3 ml of conc. HCl for 5 h: 80 mg (15.6%) of 2-methoxy-2-(α -methoxy-*p*-chlorobenzyl)acenaphthenone (4b), m.p. 193–194°, as colourless prisms and 290 mg (71%) of *p*-chloroaniline were obtained. (Found: C, 71.76; H, 4.83. Calc. for $\text{C}_{23}\text{H}_{19}\text{O}_3\text{Cl}$: C, 71.49; H, 4.82%). ν_{CO} 1732 cm^{-1} . δ ppm (CCl_4): 2.87, 3.0 (each 3H, s, CH_3), 4.74 (1H, s, \rightarrow CH), 6.5–8.2 (10H, m, aromatic protons).

Hydrolysis of 3d. A solution of 1.0 g of 3d in 20 ml conc. HCl was stirred at room temp for 4 h. The precipitate was recrystallized from petrol (b.p. 42–60°) to give 220 mg (50.5%) of acenaphthenequinone mono-*p*-toluidine (5), m.p. 189–191°, as orange yellow prisms. This compound was identical with the authentic sample prepared from acenaphthenequinone and *p*-toluidine.

The filtrate was extracted with Et₂O and the extract evaporated to leave 90 mg (20%) of *p*-chlorobenzaldehyde. The aqueous layer was neutralized with Na₂CO₃ aq. to give 50 mg (29.2%) of *p*-toluidine and trace amounts of anil 2d.

When a solution of 300 mg of 3d in 18 ml of EtOH was refluxed with 2 ml conc. HCl for 2 h, 20 mg (11%) of 4a, 60 mg (58%) of *p*-toluidine and 10 mg of acenaphthenequinone were obtained.

Reaction of 6 with 2b. A solution of 1.46 g (10 mmole) of 6 and 1.81 g (10 mmole) of 2b in 30 ml of benzene was refluxed with 80 mg of BF₃·Et₂O for 1 h. After concentration to 15 ml, the resulting solution was chromatographed on silica gel using benzene as eluent to give crystals. Recrystallization from petrol afforded 0.47 g (15.8%) of *α*-anilinobenzylidenacetophenone (7b-1), m.p. 84–86°, as yellow grains. (Found: C, 84.06; H, 5.60; N, 4.65. Calc. for C₂₁H₁₇NO: C, 84.06; H, 5.72; N, 4.68%). ν_{CO} 1625 cm⁻¹. *m/e* (rel. intensity): 299 (M⁺, 27), 194 (12), 193 (14), 105 (84), 77 (100). δ ppm (CDCl₃): 12.2 (1H, NH).

Similarly, reaction of 2.72 g of 6 with 3.75 g of 2a in the presence of 160 mg of BF₃·Et₂O in 60 ml of benzene afforded 1.78 g (26%) of *α*-*p*-chloroanilino-*p*-chlorobenzylidenacetophenone (7a-1), m.p. 160–161°, as yellow prisms. (Found: C, 68.84; H, 3.99; N, 3.67. Calc. for C₂₁H₁₅NOCl₂: C, 68.48; H, 4.08; N, 3.80%). ν_{CO} 1623 cm⁻¹. *m/e* (rel. intensity): 371 (10.6), 370 (21), 369 (55), 368 (76), 367 (89), 366 (92), 227 (42), 105 (100), 77 (98).

Preparation of β -anilinobenzylidenacetophenone (7b-2). A mixture of 1.12 g (5 mmole) of dibenzoylmethane and 0.58 g (6.2 mmole) of freshly distilled aniline was refluxed for 9 h. After cooling, the mixture was extracted with hot petrol (50 ml \times 2) and the extract evaporated. The residue was chromatographed on alumina (benzene) to afford 820 mg (55%) of 7b-2, m.p. 102–103° (lit.¹⁰ m.p. 102–103°).

Reaction of 1 with 8a. A solution of 1.0 g (5 mmole) of 1 and 1.0 g (5 mmole) of 8a in benzene (30 ml) was refluxed with 80 mg of BF₃·Et₂O for 8 h and concentrated to 15 ml. The resulting solution was chromatographed on alumina (benzene) to give crystals, which on recrystallization from EtOH afforded 730 mg (41%) of 1,2-diphenylacenaphtho[1,2-*b*]pyrrole (10a), m.p. 183–184°, as yellow prisms. (Found: C, 91.09; H, 4.73; N, 3.80. Calc. for C₂₀H₁₁N: C,

90.93; H, 4.99; N, 4.08%). δ ppm (CDCl₃): 6.7 (1H, s), 7.1–7.7 (16H, m, aromatic protons). *m/e*: 343 (M⁺).

Similarly, reaction of 1 with 8b in the presence of BF₃·Et₂O afforded 23% yield of 1,2-diphenyl-3-methylacenaphtho[1,2-*b*]pyrrole (10b), m.p. 152–153°, as orange needles (from EtOH). (Found: C, 90.70; H, 5.30; N, 3.66. Calc. for C₂₇H₁₉N: C, 90.72; H, 5.36; N, 3.92%). δ ppm (CDCl₃): 2.42 (3H, s, CH₃), 7.1–7.7 (16H, m, aromatic protons). *m/e*: 357 (M⁺).

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